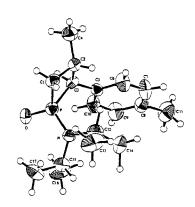
## **Index Abstracts**

The phosphenium cation [i-Pr<sub>2</sub>N-P-CI]<sup>+</sup> (1) selectively inserts into the less hindered ring C—C bond linking to the phenyl group of trans-2-methyl-1-(4-substituted)phenylcyclopropane(2) to form 1-diisopropylamino-2-(4-substituted) phenyl-3-methyl-phospheta ne-I-oxides (3). In 3c, the 1-diisopropylamino group is unexpectedly cis to 2-(p-methyl)phenyl group and the dihedral angle between the planes  $C_{(1)}$ -P- $C_{(3)}$  and  $C_{(1)}$ - $C_{(2)}$ - $C_{(3)}$  is 15.09°, which is the least in the found values.

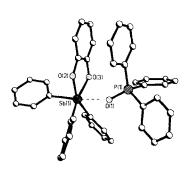


Tian-Bao Huang, Ke Wang, Ling-Fei Liu and Fa-Hu He

Insertion of ((Di-iso-propyl) amino)(chloro)phosphenium Cation into Phenylcyclopropane Ring: Stereochemistry, Regioselectivity and Mechanism

5 - 14

Six new complexes of hexa-coordinated antimony were prepared by using the "one pot" oxidative method in the reactions of trimethyl(triethyl or triphenyl)antimony with triphenylphosphine and pyrocatechol or ethylene glycol in the presence of 2 moles tert-butylhydroperoxide as well as in the reactions of triphenylantimony and pyrocatechol with tertbutylhydroperoxide in the presence of dimethylsuphuroxide or piridine oxide as monodentate ligands. The structure of (C<sub>6</sub>H<sub>4</sub>O<sub>2</sub>)SbPh<sub>3</sub>(OPPh<sub>3</sub>) was determined by X-ray diffraction method.



V. A. Dodonov, A. Yu. Fedorov, G. K. Fukin, S. N. Zaburdyaeva, L. N. Zakharov and A. V. Ignatenko

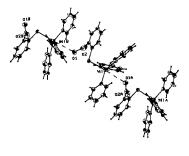
Synthesis and Structural Characterization of Some Complexes of Hexa-Coordinated Antimony

15-22

Short non-bonded intramolecular S-O interactions occur in 2,4- $(O_2N)_2C_6H_3SX$  (2:  $X=NMe_2$ ; 3:  $X=C_6H_4NO_2$ -2), but are absent in Ph<sub>3</sub> MSC<sub>6</sub>H<sub>4</sub>NO<sub>2</sub>-2 (4: M=Ge; 5: M=Sn; 6: M=Pb); however weak intermolecular M-O bonding is present in 5 and 6. The geometries at the metal centres in 5 and 6 are between trigonal bipyramidal and tetrahedral; 4 is essentially tetrahedral.

The molecular structures of  $[(^iBu)_2]$  Al $\{\mu$ -OC(H)(Ph) CH(Me)N(H)Me $\}$ ]<sub>2</sub> (1) and ( $^iBu)_2$ Al $[OC(CH_2Ph)(Ph)CH (Me)CH_2NMe_2]$  (2) have been determined by X-ray crystallography. Compound 2 is monomeric with a six membered chelate heterocyclic rings. In contrast, compound 1 exists as a bridged dimer. The formation of monomeric chelate compounds for compound 2, rather than alkoxide bridged dimers as is found for 1 is found to be due to the steric bulk of the aluminum alkyl and substitution at the ligand's  $\alpha$ -carbon.

Hydroalumination of H<sub>2</sub>C=CHCH<sub>2</sub> SMe with  $[({}^{t}Bu)_{2}Al(\mu-H)]_{3}$  yields the monomeric thioether compound (<sup>t</sup>Bu), Al(CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>SMe). The molecular structure  $(^{t}Bu)_{2}$ Al(CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>SMe) consists of discreet monomers in which the thioether-alkyl ligand acts as a chelate ligand with an Al-S interaction [2.511(2) Å] that is significantly shorter than observed for other intramolecular coordination compounds, 2.78-2.95 Å. The coordination about aluminum is highly distorted from tetrahedral and is therefore best described as capped trigonal planar or trigonal bipyramidal with a vacant coordination site.



John H. Aupers, Philip J. Cox, Solange M. S. V. Doidge-Harrison, R. Alan Howie, John N. Low and James L. Wardell

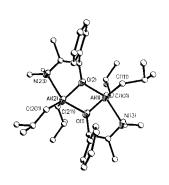
Non-bonded Interactions Involving Oxygen in 2-Nitrobenzenethiolato Compounds: Structures of  $2, 4-(O_2N)_2C_6$   $H_3SX$  ( $X = NMe_2$  or  $C_6H_4NO_2$ -2) and  $Ph_3MSC_6H_4NO_2$ -2 (M = Ge, Sn or Pb)

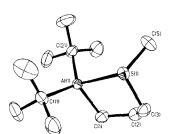
## 23-42

C. Niamh McMahon, Simon G. Bott and Andrew R. Barron

Steric Effects in Aluminum Compounds Containing Monoanionic Potentially Bidentate Ligands: Effect of the Steric Bulk at the  $\alpha$ -Carbon

43-51





Julie A. Francis, Simon G. Bott and Andrew R. Barron

Hydroalumination of  $H_2C$ = CHC $H_2SMe$ : Synthesis and Molecular Structure of  ${}^{(t}Bu)_2Al(CH_2CH_2CH_2SMe)$ 

53-57

Ab initio MO calculations indicate that TeS<sub>7</sub>, 1,2-Te<sub>2</sub>S<sub>6</sub>, 1,2-TeSeS<sub>6</sub>, and 1, 2, 8-TeSe<sub>2</sub>S<sub>5</sub> are the most stable tellurium-containing chalcogen rings. These species are also observed in the <sup>125</sup>Te NMR spectra of products obtained in the reaction of  $[Ti(MeC_5H_4)_2 (Te_xE_{4-x})_2Ti(MeC_5H_4)_2]$  (E = S, Se) with chalcogen chlorides or in the reaction of  $(Me_3Si)_2Te$  with ClSeS<sub>5</sub>-SeCl<sub>2</sub>.

Jari Taavitsainen and Risto S. Laitinen

Tellurium-Containing Chalcogen Rings

59-67